TOTAL SYNTHESIS OF DEOXYVERNOLEPIN

Paul A Grieco^{1*}, José A Noguez, and Yukio Masaki Department of Chemistry, University of Pittsburgh Pittsburgh, Pennsylvania 15260

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To date model studies directed toward the total synthesis of the sesquiterpene dilactone vernolepin(1)² have concentrated on the synthesis of the novel <u>cis</u>-fused α -methylene- δ -valerolactone AB-ring system³⁻⁷ with no attention being devoted to either relevant functionality in the B-ring or to construction of bis- α -methylene dilactone systems. An important simplification in the design of a total synthesis is possible if it can be assumed that the technology which we and others have developed for the α -methylenation of lactones can be applied in a concurrent fashion for the simultaneous introduction of both α -methylene units of vernolepin(e g 2 \rightarrow 1) Furthermore it would be useful if this operation could be postponed for the last

$$\begin{array}{c|c} OH & & OH \\ \hline \\ CH_2 & & CH_2 \\ \hline \\ CH_2 & & O \end{array}$$

stage of the synthesis Our results described below indicate that such planning is realistic. We demonstrate for the first time the feasibility of bis- α -methylenation in the context of the total synthesis of deoxyvernolepin (3), the most functionalized synthetic analog of vernolepin thus far reported

Initially our attention focused on the preparation of decalone 4 from the ketal olefin 6^9 Hydroboration of the methyl ether of octalone 6 followed by Collins oxidation afforded ketone 7^{11} Introduction of the acetic acid side chain at C-7(steroid numbering) in 7 was accomplished in a straight forward manner—a) carbomethoxylation, b) alkylation with methyl bromoacetate, and c) decarboxylation using barium hydroxide Reduction(sodium/isopropanol) and esterification of keto acid 8(mp 143-144°) provided in 55 % overall yield the crystalline hydroxy ester 9, mp 94-95° Acetylation and deketalization of 9 generated the desired decalone 4, mp 120° nmr 11a (CDCl₃) δ 2 00(s, 3H, CH₃CO-), 3 34(s, 3H, CH₃O-), 3 53(ABq, 2H, -CH₂O-, J = 9 Hz, $\Delta \nu_{AB} = 15$ 6 Hz), 3 59(s, 3H, CH₃OCO-), 4.75(t, 1H, J = 10 Hz, -CHO-)

With the stereochemical and synthetic aspects of this sequence for the formation of 4 assured, attention was turned to the transformation of 4 to the required tricyclic dilactone 5. Formation of hydroxy ester 11 \underline{via} specific cleavage of the C-2, C-3 carbon-carbon bond was accomplished by a) conversion of 4 to the Δ^2 -enol acetate 10, mp 119-120°, b) ozonolysis with a reductive workup, and c) esterification. Conversion of 11 to its corresponding mesylate followed by treatment with boron tribromide at -78° provided the valerolactone 12. Treatment of 12 with o-mitrophenylselenium anion afforded selenide 13 which upon oxidation-elimination 13 , acetate cleavage, and lactonization was smoothly converted to the desired tricvolic dilactone 5 in (CHCl₃) 5.59, 5.75, 6.10 μ , nmr 11a (CDCl₃) δ 3.98(t, J = 10 Hz, 1H), 4.78(ABq, 2H, J = 11 Hz, $\Delta\nu_{AB}$ = 30.0 Hz), 5.3 - 5.8(ABC pattern, 3H, -CH=CH₂)

 α -Hydroxymethylation ¹⁴ of dilactone 5 in tetrahydrofuran containing 10 % hexamethylphosphoramide gave a bis- α -hydroxymethyl derivative which when subjected to mesylation and β -elimination in pyridine afforded in 32 % overall yield from 5, crystalline deoxyvernolepin(3), mp 169-170° ir (CHCl₃) 5 65, 5 82 μ , nmr ^{10a} (CDCl₃) δ 6 74(s, 1H), 6 20(d, J = 2 5 Hz, 1H), 5 95(s, 1H), 5 53(d, J = 2 5 Hz, 1H), 5 3 - 5 8 (typical vinyl pattern, 8 lines, 3H), 4 43(ABq, 2H, J = 12 Hz, $\Delta\nu_{AB}$ = 81 2 Hz), 4 00(t, 1H, J = 11 Hz) Acknowledgements This investigation was supported by a Public Health Service Research Grant(CA 13689) from the National Cancer Institute, the National Institutes of Health NMR Facility for Biomedical Studies (RR-00297), and in part by Eli Lilly and Co We thank Mr. C. S Pogonowski and Ms D Boxler for experimental assistance during the preparation of decalone 7 One of us(J, A. N.) acknowledges a postdoctoral fellowship from the Umiversidad Nacional Autonoma de Mexico

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a, NaH, MeI, THF, b, BH $_3$ THF, OH $_1$, H $_2$ O $_2$, c, CrO $_3$ 2Py, CH $_2$ CI $_2$, d, NaH, (MeO) $_2$ CO, dioxane, e, NaH, BrCH $_2$ COOMe, dioxane, f, Ba(OH) $_2$ 8H $_2$ O, EtOH, reflux, g, Na, 1-PrOH, h, CH $_2$ N $_2$, Et $_2$ O, 1, (CH $_3$ CO) $_2$ O, Py, 1, 5 % HCl, THF, k, CH $_2$ =C(OAc)CH $_3$, TsOH, 1, O $_3$, MeOH CH $_2$ CI $_2$ (2 3), m, BH $_4$ $_1$, OH $_1$, n, MsCl, Py, 0°, p, 0-O $_2$ NC $_6$ H $_4$ SeCN, BH $_4$ $_1$, DMF, rt, o, BBr $_3$, CH $_2$ CI $_2$, -78°, q, 50 % H $_2$ O $_2$, THF, r, K $_2$ CO $_3$, MeOH, s, TsOH, C $_6$ H $_6$, reflux

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